

Geochemical interpretation of sequential extraction results from the Ajka Bauxite-Residue

FERENC MÓRICZ

Institute of Mineralogy and Geology, University of Miskolc, Hungary

2nd REEBAUX workshop

2020. October 16-17.

Content



- Sequential extraction methods in general
- Overview of the different types of sequential extraction methods
- Presentation of the applied method
- Characterization of the chosen samples
- Interpretation of the result from the analysis
- Conclusion

Sequential extraction methods in general

- Nowadays, it is one of the most commonly used geochemical analysis near the pH dependent test, although it is more expensive.
- Sequential extraction method was established several decades ago in order to evaluate trace element availability (originally) in soil.
- By time it proved the applicability in case of rocks as well.
- It is a perfect geochemical tool to detect which is that mineral phase or group where the searched element is bounded.
- From the first one, more generations of the test are invented.

Sequential extraction methods

➤ **Tessier type 5 steps (1979)**

- Exchangeable fraction
- Bound to carbonates
- Bound to Fe-Mn oxides
- Bound to organics matter
- Residual

➤ **BCR type 3 steps (1993) /by Standards, Measurement and Testing Programme; formerly BCR/**

- Exchangeable fraction
- Metals bound to Fe-Mn oxides
- Metals bound to organics matter and sulphides
- Residual

Sequential extraction methods

➤ ***Dold type 7 steps (2001)***

- Water soluble fraction
- Exchangeable fraction
- Fe^{III} oxyhydroxides
- Fe^{III} oxides
- Organics and secondary (Cu) sulphides
- Primary sulphides
- Residual

Sequential extraction methods

➤ *Tessier type (1979) method modified by Gu et al. (2018)*

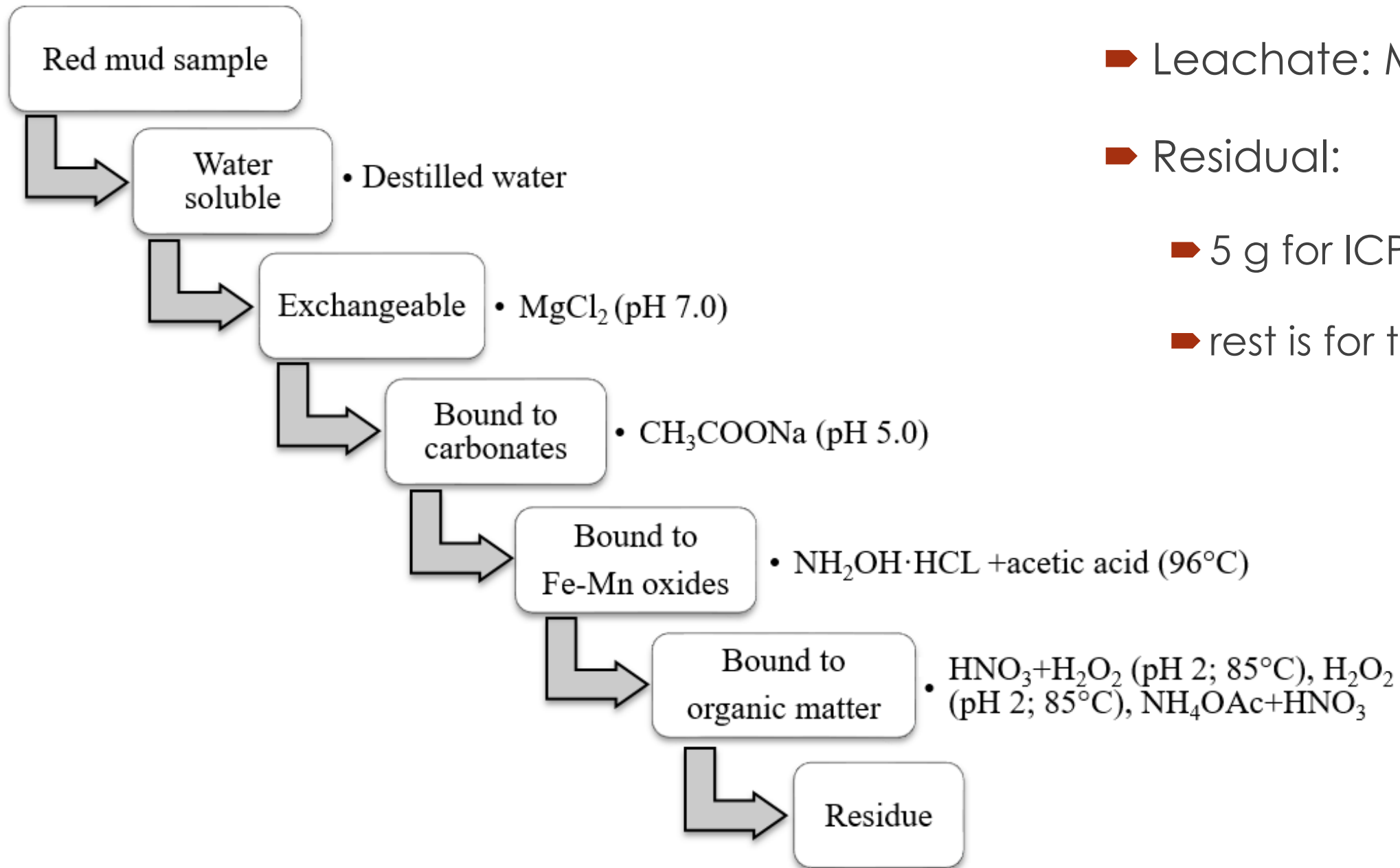
- Water soluble fraction
- Exchangeable fraction
- Bound to carbonates
- Bound to Fe-Mn oxides
- Bound to organics
- Residual

Gu, H., Wang, N., Hargreaves, J. S. J. (2018): **Sequential extraction of valuable trace elements from Bayer process-derived waste red mud samples.** *Journal of Sustainable Metallurgy*, Vol. 4, pp. 147-154.

Our choice: Gu et al. (2018) method

- **But why?**
- It is specified for red mud (pH 13-14), as the residual of the Bayer process.
- It has good mineral phase separation because of:
 - the large number of the steps
 - the selectively chosen chemicals let better detection of the mineral groups
 - compare to „old” Tessier or BCR method, in them some steps leach out more than one mineral phase in the same step
 - Dold method has the same separation but it is invented for Cu minerals
- Key answers: 1, good separation of the phases; 2, specified for red mud

The applied Gu et al. (2018) method



➤ Leachate: MP-AES

➤ Residual:

➤ 5 g for ICP-ES and ICP-MS

➤ rest is for the next step

Method has already chosen, but which sample(s) will be analyzed from the 41 species?

- ▶ Cannot forget: 1 sample = 5-times leaching process (time consuming), 5 leachate and 5 residual material.
- ▶ Based on the chemical and mineralogical results, the cell VIII. was chosen in a vertical set of the top 5 species, which mean samples from depth of 1, 2, 3, 4 and 5 m.
- ▶ 20-times test size enlargement was applied, as instead of 3 g, the starting amount was modified to 60 g of the sample.
- ▶ The same enlargement was used also for the amounts to the reagents.

Mineralogical composition of the chosen samples

- Based on these mineralogical results, the Gu et al. (2018) method can be a good choice.
- Theoretically there should have no overlapping of dissolution of the similar mineral group in the same step.

Sample *		50	51	52	53	54
Sampling depth (m)		1	2	3	4	5
Hematite	Fe ₂ O ₃	35.2	34.8	34.9	35.6	37
Cancrinite	Na ₆ Ca ₂ Al ₆ Si ₆ O ₂₄ (CO ₃) ₂	17.1	9.1	19.9	22.2	18.4
Gibbsite	Al(OH) ₃	5.1	3.0	1.7	2.2	1.2
Calcite	CaCO ₃	6.6	12.0	7.5	10.3	9.6
Dolomite	CaMg(CO ₃) ₂	0.8	2.0	0.9	0.0	1.4
Katoite	Ca ₃ Al ₂ (SiO ₄) _{1.5} (OH) ₆	2.5	1.2	4.2	3.0	2.3
Goethite	FeOOH	16.6	19.3	12.0	3.7	7.2
Boehmite	AlOOH	1.4	2.7	1.4	0.7	2.3
Quartz	SiO ₂	0.5	0.4	0.5	0.5	0.2
Kaolinite	Al ₂ Si ₂ O ₅ (OH) ₄	1.9	0.8	0.6	2.2	0.7
Cancrinite (OH)		2.8	1.8	1.8	3.6	2.7
Amorphous		8.4	12.0	13.3	14.6	15.5
Others**		1.1	0.7	1.3	1.4	1.5

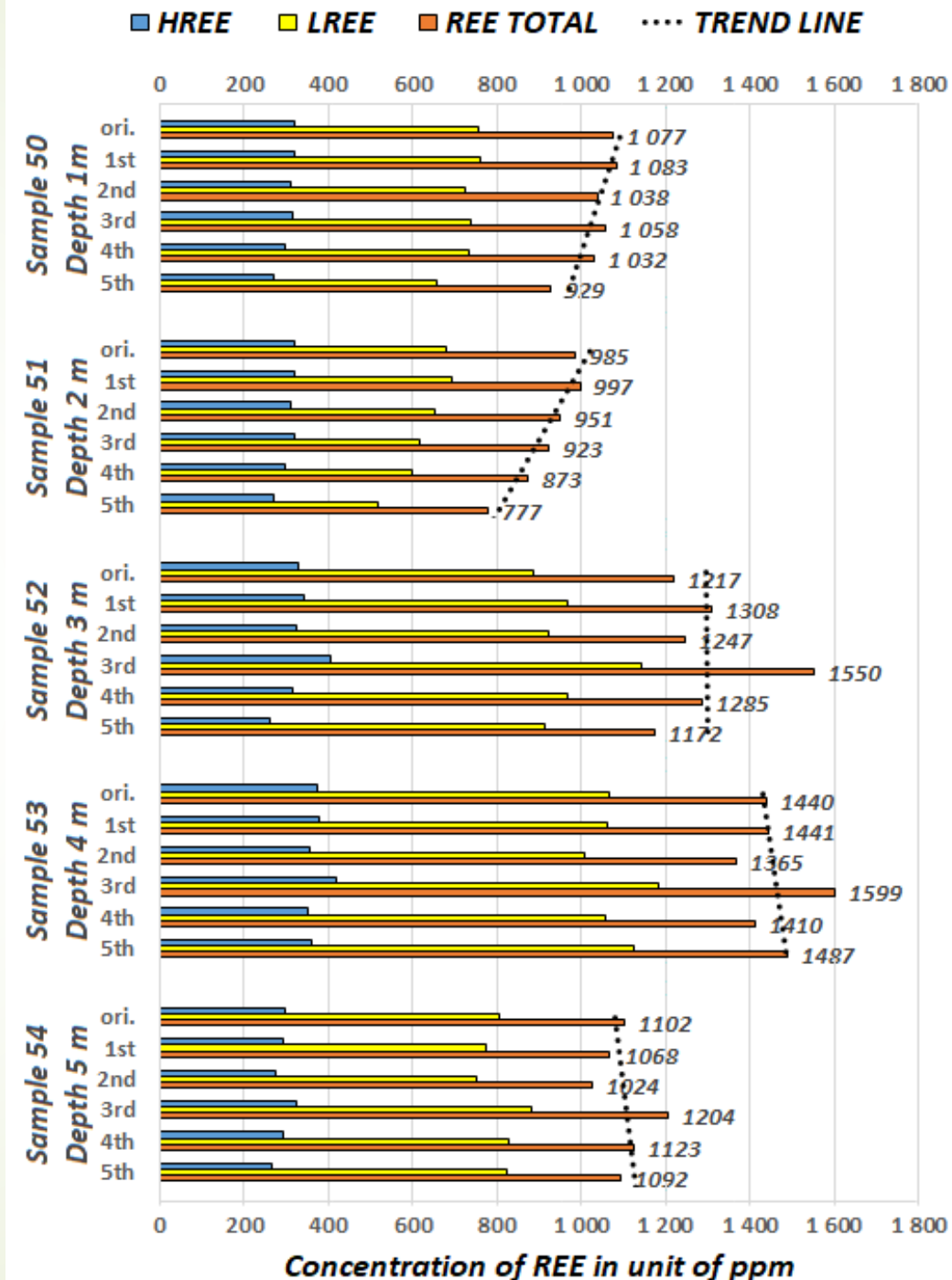
* The unit is given in m/m%.

** Total of hibschite, anatase, diaspore, manganosite and aragonite.

SE results from the residuals

Three trend line can be read out:

- Upper zone (sample 50 and 51; 1 & 2 m)
 - Decreasing REEs content => dissolving
 - REEs bound mainly to carbonates and Fe phases
- Middle zone (sample 52; 3 m)
 - Stable REEs content => partly dissolving
 - The dissolving REEs keep balance with the enrichment effect in the residual.
- Lower zone (sample 53 and 54; 4 & 5 m)
 - Increasing REEs content => higher the enrichment effect than the dissolution rate
 - REEs bound mainly to REE phosphates or silicates



Conclusion

- After literature work, proper sequential extraction method was chosen.
- From the 41 species of the red mud samples, 5 were chosen after checking the chemical and mineralogical compositions of them.
- Near the difficulties of the samples, the test was prosperous.
- From the result of the sequential extraction and from the trend lines of the changes of the REEs three type of zones were established.
- At the top, the REEs dominantly bound to the carbonate and Fe phases, while in lower they are dominantly in the phosphate and silicate phases.



Thank you for your kind attention!